

Natural and anthropogenic ethanol sources in North America and potential atmospheric impacts of ethanol fuel use

Dylan Millet, Eric Apel, Daven K. Henze, Jason Hill, Julian D Marshall, Hanwant Singh Singh, and Christopher Tessum

Environ. Sci. Technol., **Just Accepted Manuscript** • DOI: 10.1021/es300162u • Publication Date (Web): 25 Jun 2012

Downloaded from <http://pubs.acs.org> on July 2, 2012

Just Accepted

“Just Accepted” manuscripts have been peer-reviewed and accepted for publication. They are posted online prior to technical editing, formatting for publication and author proofing. The American Chemical Society provides “Just Accepted” as a free service to the research community to expedite the dissemination of scientific material as soon as possible after acceptance. “Just Accepted” manuscripts appear in full in PDF format accompanied by an HTML abstract. “Just Accepted” manuscripts have been fully peer reviewed, but should not be considered the official version of record. They are accessible to all readers and citable by the Digital Object Identifier (DOI®). “Just Accepted” is an optional service offered to authors. Therefore, the “Just Accepted” Web site may not include all articles that will be published in the journal. After a manuscript is technically edited and formatted, it will be removed from the “Just Accepted” Web site and published as an ASAP article. Note that technical editing may introduce minor changes to the manuscript text and/or graphics which could affect content, and all legal disclaimers and ethical guidelines that apply to the journal pertain. ACS cannot be held responsible for errors or consequences arising from the use of information contained in these “Just Accepted” manuscripts.



1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

Natural and anthropogenic ethanol sources in North America and potential atmospheric impacts of ethanol fuel use

Dylan B. Millet^{*1}, *Eric Apel*², *Daven K. Henze*³, *Jason Hill*¹, *Julian D. Marshall*¹, *Hanwant B. Singh*⁴,
*and Christopher W. Tessum*¹

¹University of Minnesota, Minneapolis-St. Paul, MN, USA.

²National Center for Atmospheric Research, Boulder, CO, USA.

³University of Colorado, Boulder, CO, USA.

⁴NASA Ames Research Center, Moffett Field, CA, USA.

* Corresponding author email: dbm@umn.edu

ABSTRACT: We use an ensemble of aircraft measurements with the GEOS-Chem chemical transport model to constrain present-day North American ethanol sources, and gauge potential long-range impacts of increased ethanol fuel use. We find that current ethanol emissions are underestimated by 50% in Western North America, and overestimated by a factor of two in the east. Our best estimate for year-2005 North American ethanol emissions is 670 GgC/y, with 440 GgC/y from the continental US. We apply these optimized source estimates to investigate two scenarios for increased ethanol fuel use in the US: one that assumes a complete transition from gasoline to E85 fuel, and one tied to the biofuel requirements of the US Energy Independence and Security Act (EISA). For both scenarios, increased

1 ethanol emissions lead to higher atmospheric acetaldehyde concentrations (by up to 14% during winter
2 for the All-E85 scenario) and an associated shift in reactive nitrogen partitioning reflected by an
3 increase in the PAN:NO_y ratio. The largest relative impacts occur during fall, winter and spring because
4 of large natural emissions of ethanol and other organic compounds during summer. Projected changes in
5 atmospheric PAN reflect a balance between an increased supply of peroxyacetyl radicals from
6 acetaldehyde oxidation, and the lower NO_x emissions for E85 relative to gasoline vehicles. The net
7 effect is a general PAN increase in fall through spring, and a weak decrease over the US Southeast and
8 the Atlantic Ocean during summer. Predicted NO_x concentrations decrease in surface air over North
9 America (by as much 5% in the All-E85 scenario). Downwind of North America this effect is
10 counteracted by higher NO_x export efficiency driven by increased PAN production and transport. From
11 the point of view of NO_x export from North America, the increased PAN formation associated with E85
12 fuel use thus acts to offset the associated lower NO_x emissions.
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29

30 1. INTRODUCTION

31
32 Ethanol (C₂H₅OH) is emitted to the atmosphere by vegetation, during biomass combustion, and as a
33 result of various urban and industrial processes. It is also increasingly used in the US as a biofuel mixed
34 with gasoline. In the atmosphere, ethanol is a precursor of acetaldehyde (CH₃CHO) and peroxyacetyl
35 nitrate (PAN), so that changing ethanol emissions have the potential to affect urban air pollution and
36 associated long-range transport. However, the significance of this effect will depend on the size of the
37 emission change compared to that of the existing source fluxes, which are poorly known. Here we use a
38 global 3D chemical transport model (GEOS-Chem CTM) applied to an ensemble of airborne
39 observations to derive new constraints on natural and anthropogenic ethanol sources in North America.
40 We then employ this revised source estimate as a baseline for assessing some potential large-scale
41 impacts of changing ethanol fuel use in the US.
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

Biogenic emissions from terrestrial plants are thought to be the dominant global source of

1 atmospheric ethanol¹⁻³. Ethanol is produced in plant tissues via fermentation reactions in leaves and
2 roots⁴⁻¹². It can then be oxidized to acetaldehyde and acetate and metabolized by the plant, but some of
3 the ethanol (and acetaldehyde) is released to the atmosphere via leaf stomata⁹. There is also a smaller
4 atmospheric ethanol source from dead and decaying plant matter¹³⁻¹⁵.
5
6
7
8

9
10
11 Other ethanol sources are comparatively minor and include biomass combustion¹⁶ and photochemical
12 production via cross-reactions of peroxy radicals¹⁷. Anthropogenic sources include emissions from its
13 use as a solvent, fuel, and chemical intermediate, as well as from fermentation and various industrial
14 processes¹⁸. In previous work we estimated that natural emissions from living and decaying plants
15 currently make up 90% of the global source of atmospheric ethanol².
16
17
18
19
20
21
22
23
24
25

26 Once in the atmosphere, photochemical oxidation by OH is the main ethanol sink, occurring on a
27 timescale of ~4 days¹⁹. Ethanol is also soluble in water and can be removed by wet and dry deposition,
28 which have been estimated to account for 23-35% of the total global sink^{2,3}. Ethanol oxidation proceeds
29 as:
30
31
32
33
34



38
39
40
41
42
43 with reaction mainly (90%) via channel (1b). The alkoxy and α -hydroxyalkyl radicals produced from
44 channels (1a) and (1b) go on to react with O₂ to produce acetaldehyde + HO₂^{19,20}, so that overall
45 acetaldehyde is produced with ~95% yield. Acetaldehyde is classified as a hazardous air pollutant by the
46 US EPA²¹, and its subsequent oxidation can lead to production of ozone (O₃) and PAN, thus affecting
47 the partitioning and fate of reactive nitrogen (NO_y). The β -hydroxyalkyl radical produced via channel
48 (1c) predominantly goes on to produce glycoaldehyde and formaldehyde^{20,22,23}.
49
50
51
52
53
54
55
56
57
58
59
60

1 Ground-based ethanol measurements to constrain biogenic and anthropogenic sources are sparse, and
2 as a result emission fluxes are uncertain. Aircraft measurements during the Intercontinental Transport
3 Experiment, Phase A (INTEX-A, Jul-Aug 2004) and Phase B (INTEX-B, Apr-May 2006), and the
4 Megacity Initiative: Local and Global Research Observations (MILAGRO, Mar 2006) featured
5 extensive vertical profiling and boundary layer measurements over North America and the adjacent
6 oceans. Our objective here is to apply these data to derive new information on North American ethanol
7 emissions, and to use that information to evaluate some atmospheric impacts of potential future changes
8 in ethanol fuel use, focusing on the role of ethanol as a precursor of acetaldehyde and PAN, and the
9 associated effects on NO_y partitioning and export.
10
11
12
13
14
15
16
17
18
19
20
21
22
23

24 **2. AIRCRAFT MEASUREMENTS**

25
26 Figure 1 shows flight tracks for the aircraft campaigns used here. INTEX-A²⁴ took place during July-
27 August 2004, with a focus on understanding tropospheric composition over North America, the outflow
28 of pollution from North America, and its chemical evolution downwind. MILAGRO²⁵ and INTEX-B²⁶
29 were aimed at understanding air pollution transport from Mexico City and Asia and the associated
30 climatic effects. MILAGRO was carried out over Mexico, the US Gulf Coast, and the Gulf of Mexico
31 region during March 2006, while INTEX-B took place in April-May 2006 over the US West Coast and
32 the Pacific Ocean. Details on the ethanol measurements for each campaign are provided in the
33 Supporting Information (SI).
34
35
36
37
38
39
40
41
42
43
44
45
46
47

48 **3. MODELING FRAMEWORK**

49
50 We apply here the ethanol simulation implemented in GEOS-Chem by Millet et al.² GEOS-Chem
51 (v8, <http://www.geos-chem.org>) is a global Eulerian chemical transport model driven by assimilated
52 meteorological observations from the NASA Goddard Earth Observing System (GEOS-5.2.0²⁷). The
53 meteorological fields have 0.5°×0.667° horizontal resolution and 72 vertical layers. For our work here
54
55
56
57
58
59
60

1 we reduce the spatial resolution to $2^{\circ} \times 2.5^{\circ}$ and 47 layers, of which 14 are below 2 km altitude, and use a
2
3 15-minute transport timestep. We also use GEOS-4 data²⁸ to test the sensitivity of our results to the
4
5 input meteorological data. Changes to the model physics, native horizontal and vertical resolution
6
7 ($1^{\circ} \times 1.25^{\circ}$ and 55 levels for GEOS-4), and observing system between GEOS-4 and GEOS-5 lead to
8
9 significant differences in convection, boundary layer mixing, and wind speeds between the two
10
11 datasets^{29,30}. A detailed description of the GEOS-4 and GEOS-5 models, and the associated
12
13 meteorological products, is provided elsewhere^{27,28}. Simulations shown here are for 2006, and follow a
14
15 1-year model spinup.
16
17
18
19
20

21
22 Figure S1 shows North American ethanol sources as simulated by GEOS-Chem, and taken as a priori
23
24 for our source optimization based on the aircraft observations. Biogenic ethanol emissions are computed
25
26 online in GEOS-Chem using the Model of Emissions of Gases and Aerosols from Nature
27
28 (MEGANv2.1)^{2,31} as described in the SI. Fluxes are estimated for each model grid square as a sum of
29
30 contributions from four plant functional types (PFTs - broadleaf trees, needleleaf trees, shrubs, and
31
32 herbaceous plants [crops + grasses]), and the resulting North American fluxes are shown in Figure S1.
33
34 Total contributions from broadleaf trees, needleleaf trees, and shrubs are similar (0.21-0.25 TgC/y), but
35
36 each has a distinct spatial pattern reflecting that of the corresponding PFT. Predicted ethanol emissions
37
38 from herbaceous plants (grasses and crops) are lower (0.06 TgC/y) and concentrated in the Central US.
39
40 There have been only a few direct measurements of ethanol emissions from vegetation^{10,14,32} and as a
41
42 result the MEGANv2.1 flux estimates are quite uncertain, providing part of the motivation for this work.
43
44
45
46
47
48
49

50
51 A priori anthropogenic emissions over North America, including emissions from ethanol-fueled
52
53 vehicles, are from the US EPA's National Emission Inventory (NEI) for 2005³³. As shown in Figure S1,
54
55 annual anthropogenic emissions for North America total 0.22 TgC/y in the model, or 30% of the
56
57 simulated biogenic source. Other minor ethanol sources included in the model (but not plotted in Figure
58
59
60

1 S1) include biomass combustion and photochemical production (0.001 and 0.004 TgC/y, respectively
2 over North America), calculated as in Millet et al.²
3
4
5
6

7 Photochemical destruction of ethanol by OH is computed in the model based on a rate constant¹⁹ $k =$
8 $3.0 \times 10^{-12} \exp(20/T)$ applied to global OH fields archived from a full-chemistry GEOS-Chem simulation.
9
10 Other ethanol sinks include wet and dry deposition. Modeled wet deposition in GEOS-Chem includes
11 scavenging in wet convective updrafts, rainout and washout from convective anvils and large-scale
12 precipitation^{34,35}, while dry deposition follows a resistance-in-series model^{36,37}. Modeled photochemical
13 and depositional sinks for ethanol are plotted in Figure S1 over North America, and together result in a
14 global annual average lifetime of 3 days.
15
16
17
18
19
20
21
22
23
24
25

26 Figure S2 shows the sensitivity of the simulated ethanol concentrations along the aircraft flight tracks
27 to the various North American ethanol sources described above. With the exceptions of photochemical
28 production and biomass burning, we see from comparing Figures S2 and S1 that the combination of
29 airborne datasets used here provides comprehensive coverage with respect to the distribution of North
30 American source influences.
31
32
33
34
35
36
37
38
39

40 **4. OPTIMIZED NORTH AMERICAN ETHANOL SOURCES**

41
42
43 Figure 2 compares boundary layer (taken here as $P > 750$ hPa) ethanol measurements from the flight
44 campaigns described in Section 2 to those simulated by GEOS-Chem along the flight track at the time
45 of measurement. There are clear, and geographically specific, model biases. Most notably, simulated
46 concentrations are too high in the Eastern US and too low in the Central and Western US. Here we
47 apply these comparisons to derive the emission fluxes that are most consistent with the observational
48 constraints using a Bayesian optimization approach.
49
50
51
52
53
54
55
56
57
58
59
60

Applying Bayes' theorem and assuming Gaussian error distributions, the optimal set of ethanol emissions is that which minimizes the cost function $J(\mathbf{x})$ ^{38,39}:

$$J(\mathbf{x}) = (\mathbf{x} - \mathbf{x}_a)^T \mathbf{S}_a^{-1} (\mathbf{x} - \mathbf{x}_a) + (\mathbf{F}(\mathbf{x}) - \mathbf{y})^T \mathbf{S}_\Sigma^{-1} (\mathbf{F}(\mathbf{x}) - \mathbf{y}) \quad (3)$$

Here \mathbf{y} is the vector of aircraft observations and $\mathbf{F}(\mathbf{x})$ the corresponding model values. The vector \mathbf{x} represents the sources being optimized, with \mathbf{x}_a their a priori (initial guess) values. \mathbf{S}_a and \mathbf{S}_Σ are the a priori and observational error covariance matrices. The minimum $J(\mathbf{x})$ thus defines the set of ethanol fluxes that minimizes the (error-weighted) mismatch between the aircraft data and the model, plus the (error-weighted) mismatch between the derived fluxes and their a priori values.

Previous work has shown that current models exhibit a severe low bias relative to ethanol measurements in remote areas and in the free troposphere³. Naik et al.³ speculated that this might reflect a missing secondary source, or possibly measurement artifacts manifesting at very low ambient concentrations. We therefore focus our analysis on measurements over the North American continent and in the boundary layer ($P > 750$ hPa), in order to better isolate the signal from North American surface emissions. We also remove statistical outliers (> 0.95 quantile, representing 140 of 3792 datapoints) to avoid undue influence from fresh pollution plumes that are not resolved at the resolution of the CTM.

Our initial analyses employed a state vector composed of scale factors for specific emission categories: biogenic ethanol emissions from each of the four plant functional types, and anthropogenic emissions. However, optimizing the state vector in this way did not lead to a significant reduction in the cost function, perhaps indicating that the spatial distribution of fluxes within these categories is not accurately captured in the bottom-up estimates. We therefore optimize instead a state vector composed of four regional scale factors: ethanol emissions from the Western, Central, and Eastern parts of the US plus Canada, and from Mexico (Figure S3). Emission errors (estimated at 100%) for these four broad

1 regions can be assumed uncorrelated, so that \mathbf{S}_a is diagonal. The observational error \mathbf{S}_Σ includes both
2 measurement and model error. The measurement uncertainty is estimated at 20% + 20 ppt (see SI). We
3 further assume a 20% error in the forward model, representing the limit in the ability of the model to
4 capture atmospheric gradients even if the emissions were perfectly correct. We apply these uncertainties
5 independently to the measured and simulated concentrations, and add the results in quadrature to obtain
6 the overall observing system error \mathbf{S}_Σ . Errors are assumed uncorrelated at the $2^\circ \times 2.5^\circ$ model resolution
7 so that \mathbf{S}_Σ is diagonal. Later we carry out a sensitivity analysis to assess the extent to which our findings
8 depend on the specific construction of \mathbf{S}_a and \mathbf{S}_Σ .
9
10
11
12
13
14
15
16
17
18
19
20

21 Table 1 shows the a posteriori correction factors for ethanol emissions from the four regions. We find
22 a large model underestimate of ethanol emissions for the Western region, and a large overestimate for
23 the Eastern region as well as for Mexico. Including all aircraft datasets as a single ensemble in the
24 inversion, and solving for time-invariant scale factors, we infer a 53% model underestimate for the
25 Western region and overestimates of a factor of 2 and 1.7 for the Eastern region and Mexico,
26 respectively (a posteriori scale factors of 0.47 and 0.60). This west/east discrepancy is similar to recent
27 findings for methanol based on satellite data⁴⁰. There appears to be a persistent MEGAN underestimate
28 of certain oxygenated volatile organic compound (VOC) emissions for plant types prevalent in the
29 Western US compared to the Eastern US. Recent in-situ measurements in Central California imply very
30 large methanol and ethanol emissions in that region (J. de Gouw, personal communication), apparently
31 related to agriculture^{41,42}, and this is probably also a contributing factor. The base-case inversion
32 indicates a 1.7× model overestimate for the Central US + Canada; however, based on the sensitivity
33 analyses described below, emissions from this region are less well constrained by the aircraft
34 observations.
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56

57 Figure S4 compares the a posteriori boundary layer ethanol concentrations with the corresponding
58
59
60

1 aircraft observations. As we see, the high model bias in the Eastern US is removed, while the low bias in
2 the Western US is reduced but not completely eliminated. As we discuss in the next section, this reflects
3 the fact that the springtime, western-focused measurements indicate a larger correction to the modeled
4 Western US source than do the summertime, eastern-focused measurements. The a posteriori simulation
5 also shows evidence of a residual low bias in the North Central US; however, the sparse observational
6 coverage over this region prevents any definitive constraint.
7
8
9
10
11
12
13
14
15
16

17 We thus estimate present-day ethanol emissions from the continental US at 440 GgC/year (Table 1).
18 The corresponding emissions from North America as a whole are plotted in Fig. 3 (left column), and
19 total 670 GgC/y with a strong seasonal cycle. This can be compared to the estimated present-day
20 emissions for other non-methane VOCs, totaling 68 TgC/y (Figure S5, left column).
21
22
23
24
25
26
27

28 5. UNCERTAINTY ANALYSIS

29
30
31 Previous inverse analyses have shown that Bayesian a posteriori errors tend to underestimate the true
32 uncertainty in the solution, and that a more accurate uncertainty estimate can be obtained by repeating
33 the optimization while varying key forward model parameters and error specifications⁴³⁻⁴⁶. We follow
34 that approach here, focusing on ethanol sinks and model transport as the two main model processes
35 where inaccuracies could be conflated with an emission bias. Changing the specification of the a priori
36 and observational error covariance matrices (\mathbf{S}_a and \mathbf{S}_Σ) does not have a significant effect on the solution
37 (< 1% change in the emission estimates for doubling or halving the error estimates).
38
39
40
41
42
43
44
45
46
47
48
49

50 Table 1 shows the results from an ensemble of sensitivity runs carried out to test the robustness of our
51 optimization. Ethanol is removed from the atmosphere by reaction with OH and by deposition.
52 Increasing the deposition efficiency for ethanol by allowing for reactive uptake by vegetation
53 (increasing f_0 from 0 to 1)⁴⁷, or employing OH fields from a different model version (corresponding to a
54
55
56
57
58
59
60

1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

6.4% decrease in global average OH), results in a $\leq 5\%$ change to the inferred emissions from the Eastern US + Canada, Western US + Canada, and Mexico. Emissions from the Central US and Canada are slightly less well constrained, changing by 8-14% for these two sensitivity runs. All derived ethanol fluxes, particularly for the Central region, are more sensitive to the meteorological fields used to drive GEOS-Chem. Employing GEOS-4 rather than GEOS-5 meteorological data, with the associated changes to boundary layer mixing and ventilation²⁷⁻³⁰, modifies our a posteriori emission estimates by +3%, -14%, and +17% for the Western region, Eastern region, and Mexico, respectively, and by +43% for the Central US + Canada. However, the a posteriori cost-function for the GEOS-4 based optimization is 40% higher than for GEOS-5, reflecting a much poorer ability to match the observations with the GEOS-4 dataset.

There is also a temporal aspect to consider in this analysis. The aircraft measurements in the Western US took place during springtime while those in the east took place during summer, so part of the east-west discrepancy could reflect a bias in the modeled seasonality of emissions. To test this, we repeated the analysis separately for the springtime and summertime data. Both seasonal ensembles agree in confirming a model underestimate of the Western US source. The correction indicated by the springtime, western-focused data alone (+79%) is significantly larger than implied by the summertime data alone (+35%). This may indicate a stronger biogenic source underestimate in the model earlier versus later in the growing season, as has been seen for methanol^{48,49}. Another factor to consider is that the western-focused INTEX-B/MILAGRO data were collected in 2006, two years later than the eastern-focused INTEX-A data; increases in anthropogenic ethanol emissions between 2004-2006 thus may also contribute to the observed east-west discrepancy. When considered alone, the summer campaigns imply a 2 \times decrease in emissions from the Eastern US, nearly identical to the result using the full data ensemble. The springtime data alone do not have adequate sensitivity to constrain ethanol emissions from the Central or Eastern US (Figure S6), due to the western-focused flight strategy and prevailing

1 west-to-east winds.
2
3
4

5 For the time-invariant inversion, the a posteriori cost function is a factor of 1.5 lower than the a priori
6 cost function (Table 1), indicating a significant reduction in model bias, but also some residual model
7 error. This could partly reflect a seasonality bias as discussed above. Model errors related to the spatial
8 distribution of ethanol emissions within each of the regions used to construct the state vector probably
9 also contribute. More long-term observations are needed to resolve these issues, and to develop better
10 constraints on seasonal ethanol emissions from different landscapes. Expanded measurements of the
11 carbon isotopic signature of atmospheric ethanol would also help to better refine source estimates⁵⁰.
12
13
14
15
16
17
18
19
20
21
22
23

24 While the above sensitivity runs do not span every conceivable source of model and observational
25 error, they do serve to assess some key sources of uncertainty and their relative magnitude. In particular,
26 uncertainties associated with limited data coverage and model transport emerge as predominant error
27 sources in constraining the present-day ethanol budget. Overall, our sensitivity analyses are all within
28 17% of our best-guess optimization for the Western, Eastern, and Mexican regions. Emissions from the
29 Central part of the US and Canada are less well constrained (varying by up to 43% depending on the
30 model configuration), since none of the flight campaigns focused on this area. However, the derived
31 ethanol source for the continental US as a whole (440 GgC/y, Table 1) is highly consistent between the
32 optimizations, varying by a maximum of 6% for the different inversions. Similarly, total North
33 American emissions (estimated at 671 GgC/y) differ by at most 11% between the various sensitivity
34 runs when using the full aircraft data ensemble.
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51

52 **6. POTENTIAL LARGE-SCALE EFFECTS OF INCREASED ETHANOL FUEL USE**

53
54

55 In this section we use our new, top-down ethanol source estimates for North America as a baseline for
56 examining some potential effects of increasing US ethanol fuel use. Previous work has explored how air
57
58
59
60

1 quality impacts of a transition to ethanol fuel might affect human health among the US population^{51,52}.
2
3 Here, we focus on potential larger-scale atmospheric impacts and air pollution transport, with an
4
5 emphasis on the importance of ethanol as a precursor of acetaldehyde and PAN, and the related effects
6
7 on NO_y partitioning and export.
8
9

10
11 We investigate two scenarios, and employ 2005 as our baseline year to match the NEI-2005 emission
12
13 inventory. Total US gasoline consumption in 2005 was approximately 140 billion gallons (bg),
14
15 compared to 4.0 bg of ethanol fuel⁵³. The first scenario assumes a complete transition of US gasoline
16
17 vehicles to E85 (85% ethanol, 15% gasoline) fuel. For this scenario, current gasoline vehicle emissions
18
19 in the model are switched to the corresponding emission profiles for E85 vehicles. While today's
20
21 vehicle fleet is not capable of running entirely on E85 fuel, we apply this scenario as a diagnostic to
22
23 evaluate the potential for E85 fuel use to affect long-range atmospheric chemistry relative to a similar
24
25 vehicle fleet running mainly on gasoline.
26
27
28
29
30
31

32
33 The second scenario reflects the US Energy Independence and Security Act (EISA) of 2007, requiring
34
35 the use of 36 bg/y of biofuel by 2022. We examine here implications of meeting this requirement
36
37 entirely with ethanol fuel. Because of the lower energy density of ethanol compared to gasoline, 36 bg
38
39 of ethanol is approximately equivalent to 24 bg of gasoline⁵⁴. To meet the EISA mandate in the model
40
41 scenario, we assume all gasoline is blended with 10% ethanol, and then increase E85 use to the point
42
43 where total US ethanol consumption reaches 36 billion gallons. Accounting for the use of denaturant,
44
45 this results in consumption of 126 bg/y of E10 (at 97% the energy content of gasoline⁵⁴) and 28 bg/y of
46
47 E85 (at 71% the energy content of gasoline⁵⁴). For both the All-E85 and EISA scenarios, we consider
48
49 only changes to tailpipe emissions; our future work will quantify full life-cycle emission changes for
50
51 various biofuel production strategies.
52
53
54
55
56
57
58
59
60

1 There is a marked shift in the speciation of emitted VOCs for ethanol-based fuel versus gasoline, with
2 the dominant organic compound emitted by E85-fueled vehicles being ethanol itself⁵⁵. VOC speciation
3 profiles used here for E10 and E85 emissions are based on the EPA NEI-2005 recommendations.
4 Changes in CO, NO_x, and total VOC emissions for E85 are computed based on the statistics compiled
5 by Yanowitz and McCormick⁵⁵. CO and NO_x emissions are reduced by 13% and 14% for E85 compared
6 to gasoline, corresponding to the overall geometric mean difference between E85 vehicles and similar
7 non-flex-fuel vehicles based on EPA certification data⁵⁵. The corresponding change in total VOC
8 emissions is not statistically significant, so we keep this unchanged between E85 and gasoline vehicles.
9 For E10 emissions, reported measurements show a decrease in CO emissions compared to gasoline,
10 with measured reductions between 12-25% for specific vehicles⁵⁶. Fleet-wide, EPA's MOBILE6.2
11 model predicts an average CO decrease of 6.7-7.5% for E10. There is no consistent evidence for a
12 general decrease or increase in NO_x or total VOC emissions for E10 vehicles compared to gasoline, with
13 some studies reporting an increase and others a decrease⁵⁶. Here we assume a 7.5% decrease in CO
14 emissions for E10 relative to gasoline vehicles, with no change in NO_x or total VOC emissions.
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35

36 Figures 3 and S5 show emissions of ethanol and other VOCs for present-day and projected for the
37 All-E85 and EISA scenarios (annual emissions of ethanol, other VOCs, CO and NO_x are summarized in
38 Table S1). North American ethanol emissions increase by 1040 GgC/y (2.6×) in the All-E85 scenario,
39 and by 175 GgC/y (1.3×) in the EISA scenario. Emissions from the continental US increase by 3.4× and
40 1.4× for the two scenarios. The relative perturbation is weakest in summer because of large natural
41 emissions of ethanol (and other VOCs). For instance, in the All-E85 scenario, wintertime North
42 American ethanol emissions increase by 600%, with ethanol alone then making up 6.7% of the total
43 continental VOC source at this time of year. For the same scenario during summer, ethanol emissions
44 increase by 70%, and account for only 1.7% of the total seasonal VOC source (Fig. 3, S5).
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

Figures 4 and S7 show projected changes in atmospheric ethanol, acetaldehyde, PAN:NO_y ratio, PAN, and NO_x for the All-E85 and EISA scenarios, respectively, over and downwind of North America. In each case, we see the largest fractional change during winter. For example, acetaldehyde concentrations increase by as much as 14% during winter and by up to 5% during summer for the All-E85 scenario (but by only 2% and 1% for the EISA scenario). This acetaldehyde increase leads to a larger supply of peroxy acetyl (CH₃C(O)OO) radicals and in turn a shift in reactive nitrogen partitioning, with PAN making up a larger fraction of total NO_y (Figure 4, S7).

PAN concentration changes reflect a competition between the increased supply of peroxyacetyl radicals from ethanol oxidation and the decreased NO_x emissions for E85 relative to gasoline vehicles. The net effect in our simulations is a PAN increase of up to 6% in the All-E85 scenario (1% in the EISA scenario) during fall, winter and spring. During summer there is an ample existing supply of CH₃C(O)OO and other peroxyacyl radicals from oxidation of biogenic VOCs, so that the additional ethanol source represents a smaller perturbation, and the NO_x effect becomes more important. In fact, over the Southeast US and downwind of North America we see a weak PAN decrease during summer, driven by the lower NO_x emissions (Figure 4).

Figure 4 shows, for the All-E85 scenario, a widespread decrease in surface NO_x (as much as 5%) over North America due to the lower NO_x emissions for E85 relative to gasoline vehicles. This is accompanied by a modest regional decrease in surface ozone (of order 1%). We can expect that a NO_x decrease will increase ozone within NO_x-saturated urban cores; however, our global-scale model can not fully resolve such effects. Downwind of the continent, the lower NO_x emissions are offset by increased NO_x export efficiency associated with enhanced PAN formation. Over the Atlantic Ocean, instead of a NO_x decrease, we instead see a minor increase of up to 2% outside of the summer season (Figure 4). Only during summer, when biogenic VOC oxidation provides a strong continental source of peroxyacyl

1 radicals, is there an overall (weak) NO_x decrease downwind over the Atlantic associated with E85 use.
2
3 Thus from the point of view of the amount of NO_x exported from North America, the increased PAN
4
5 formation associated with E85 fuel use acts to offset the lower NO_x emissions. These seasonal effects
6
7 are notably weaker for the EISA scenario (Figure S7) because the PAN perturbation is smaller, and
8
9 because NO_x emissions are unchanged in the model between E10 and gasoline.
10
11
12
13
14

15 In addition to the tailpipe emissions considered here, the overall atmospheric impacts of increased
16
17 ethanol fuel use will depend on emission changes during the full life-cycle of fuel production and
18
19 distribution. Depending on the feedstocks used for ethanol production, this may also include landcover
20
21 shifts that lead to changes in biogenic emissions of ethanol and other VOCs. Our results here provide
22
23 new constraints on present-day ethanol sources from North America, and on the potential for increasing
24
25 ethanol emissions to mediate how changing NO_x emissions affect the composition of continental
26
27 outflow.
28
29
30
31
32
33
34
35

36 **ACKNOWLEDGMENT:** This work was carried out in part using computing resources at the
37
38 University of Minnesota Supercomputing Institute. We thank Alex Guenther for his helpful input.
39
40
41
42

43 **SUPPORTING INFORMATION AVAILABLE:** Additional material related to the airborne ethanol
44
45 measurements, model sources of ethanol and other VOCs, the ethanol source optimization, and
46
47 projected changes to atmospheric composition for the EISA scenario are provided in the Supporting
48
49 Information. This material is available free of charge via <http://pubs.acs.org/>.
50
51
52
53
54
55

56 57 REFERENCES 58 59 60

- 1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
- (1) Kirstine, W.V.; Galbally, I.E. The global atmospheric budget of ethanol revisited. *Atmos. Chem. Phys.* **2012**, *12*, 545–555.
- (2) Millet, D.B.; Guenther, A.; Siegel, D.A.; Nelson, N.B.; Singh, H.B.; de Gouw, J.A.; Warneke, C.; Williams, J.; Eerdeken, G.; Sinha, V.; Karl, T.; Flocke, F.; Apel, E.; Riemer, D.D.; Palmer, P.I.; Barkley, M. Global atmospheric budget of acetaldehyde: 3D model analysis and constraints from in-situ and satellite observations. *Atmos. Chem. Phys.* **2010**, *10*, 3405-3425.
- (3) Naik, V.; Fiore, A.M.; Horowitz, L.W.; Singh, H.B.; Wiedinmyer, C.; Guenther, A.; de Gouw, J.A.; Millet, D.B.; Goldan, P.D.; Kuster, W.C.; Goldstein, A. Observational constraints on the global atmospheric budget of ethanol. *Atmos. Chem. Phys.* **2010**, *10*, 5361-5370.
- (4) Cojocariu, C.; Kreuzwieser, J.; Rennenberg, H. Correlation of short-chained carbonyls emitted from *Picea abies* with physiological and environmental parameters. *New Phytol.* **2004**, *162*, 717-727.
- (5) Jardine, K.; Harley, P.; Karl, T.; Guenther, A.; Lerdau, M.; Mak, J.E. Plant physiological and environmental controls over the exchange of acetaldehyde between forest canopies and the atmosphere. *Biogeosciences* **2008**, *5*, 1559-1572.
- (6) Jardine, K.J. The Exchange of acetaldehyde between plants and the atmosphere: Stable carbon isotope and flux measurements. PhD thesis, Stony Brook University, Stony Brook, NY, available at <http://dspace.sunyconnect.suny.edu/handle/1951/45198>, 2008.
- (7) Kimmerer, T.W.; Macdonald, R.C. Acetaldehyde and ethanol biosynthesis in leaves of plants. *Plant Physiol.* **1987**, *84*, 1204-1209.
- (8) Kreuzwieser, J.; Scheerer, U.; Rennenberg, H. Metabolic origin of acetaldehyde emitted by poplar (*Populus tremula* x *P-alba*) trees. *J. Exp. Bot.* **1999**, *50*, 757-765.
- (9) Kreuzwieser, J.; Schnitzler, J.P.; Steinbrecher, R. Biosynthesis of organic compounds emitted by

1 plants. *Plant Biol.* **1999**, *1*, 149-159.

2
3
4 (10) Schade, G.W.; Goldstein, A.H. Fluxes of oxygenated volatile organic compounds from a
5 ponderosa pine plantation. *J. Geophys. Res.* **2001**, *106*, 3111-3123.
6
7

8
9 (11) Schade, G.W.; Goldstein, A.H. Plant physiological influences on the fluxes of oxygenated
10 volatile organic compounds from ponderosa pine trees. *J. Geophys. Res.* **2002**, *4082*,
11 10.1029/2001JD000532.
12
13

14
15 (12) Winters, A.J.; Adams, M.A.; Bleby, T.M.; Rennenberg, H.; Steigner, D.; Steinbrecher, R.;
16 Kreuzwieser, J. Emissions of isoprene, monoterpene and short-chained carbonyl compounds from
17 Eucalyptus spp. in southern Australia. *Atmos. Environ.* **2009**, *43*, 3035-3043.
18
19

20 (13) Isidorov, V.A.; Vinogorova, V.T.; Rafalowski, K. HS-SPME analysis of volatile organic
21 compounds of coniferous needle litter. *Atmos. Environ.* **2003**, *37*, 4645-4650.
22
23

24 (14) Kirstine, W.; Galbally, I.; Ye, Y.R.; Hooper, M. Emissions of volatile organic compounds
25 (primarily oxygenated species) from pasture. *J. Geophys. Res.* **1998**, *103*, 10605-10619.
26
27

28 (15) Warneke, C.; Karl, T.; Judmaier, H.; Hansel, A.; Jordan, A.; Lindinger, W.; Crutzen, P.J.
29 Acetone, methanol, and other partially oxidized volatile organic emissions from dead plant matter by
30 abiological processes: Significance for atmospheric HO_x chemistry. *Global Biogeochem. Cy.* **1999**, *13*,
31 9-17.
32
33

34 (16) Andreae, M.O.; Merlet, P. Emission of trace gases and aerosols from biomass burning. *Global*
35 *Biogeochem. Cy.* **2001**, *15*, 955-966.
36
37

38 (17) Villenave, E.; Lesclaux, R. Kinetics of the cross reactions of CH₃O₂ and C₂H₅O₂ radicals with
39 selected peroxy radicals. *J. Phys. Chem.* **1996**, *100*, 14372-14382.
40
41
42
43
44
45
46
47
48
49
50
51
52

- 1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
- (18) Howard, P.H., *Handbook of Environmental Fate and Exposure Data for Organic Chemicals, Volume II - Solvents*. CRC Press: New York, 1990.
- (19) Atkinson, R.; Baulch, D.L.; Cox, R.A.; Crowley, J.N.; Hampson, R.F.; Hynes, R.G.; Jenkin, M.E.; Rossi, M.J.; Troe, J. Evaluated kinetic and photochemical data for atmospheric chemistry: Volume II - gas phase reactions of organic species. *Atmos. Chem. Phys.* **2006**, *6*, 3625-4055.
- (20) Atkinson, R. Atmospheric chemistry of VOCs and NO_x. *Atmos. Environ.* **2000**, *34*, 2063-2101.
- (21) *Chemical summary for acetaldehyde*; EPA 749-F-94-003a; US Environmental Protection Agency: Washington, DC, 1994; http://www.epa.gov/chemfact/s_acetal.txt.
- (22) Jenkin, M.E.; Saunders, S.M.; Pilling, M.J. The tropospheric degradation of volatile organic compounds: A protocol for mechanism development. *Atmos. Environ.* **1997**, *31*, 81-104.
- (23) Saunders, S.M.; Jenkin, M.E.; Derwent, R.G.; Pilling, M.J. Protocol for the development of the Master Chemical Mechanism, MCM v3 (Part A): Tropospheric degradation of non-aromatic volatile organic compounds. *Atmos. Chem. Phys.* **2003**, *3*, 161-180.
- (24) Singh, H.B.; Brune, W.H.; Crawford, J.H.; Jacob, D.J.; Russell, P.B. Overview of the summer 2004 intercontinental chemical transport experiment - North America (INTEX-A). *J. Geophys. Res.* **2006**, *111*, D24S01, doi:10.1029/2006JD007905.
- (25) Molina, L.T.; Madronich, S.; Gaffney, J.S.; Apel, E.; de Foy, B.; Fast, J.; Ferrare, R.; Herndon, S.; Jimenez, J.L.; Lamb, B.; Osornio-Vargas, A.R.; Russell, P.; Schauer, J.J.; Stevens, P.S.; Volkamer, R.; Zavala, M. An overview of the MILAGRO 2006 Campaign: Mexico City emissions and their transport and transformation. *Atmos. Chem. Phys.* **2010**, *10*, 8697-8760.
- (26) Singh, H.B.; Brune, W.H.; Crawford, J.H.; Flocke, F.; Jacob, D.J. Chemistry and transport of pollution over the Gulf of Mexico and the Pacific: Spring 2006 INTEX-B campaign overview and first

1 results. *Atmos. Chem. Phys.* **2009**, *9*, 2301-2318.

2
3 (27) *The GEOS-5 Data Assimilation System - Documentation of Versions 5.0.1, 5.1.0, and 5.2.0*;
4 NASA Global Modeling and Assimilation Office: Greenbelt, MD, 2008;
5
6 <http://gmao.gsfc.nasa.gov/pubs/docs/Rienecker369.pdf>.
7
8

9
10 (28) *Documentation and Validation of the Goddard Earth Observing System (GEOS) Data*
11 *Assimilation System - Version 4*; NASA Global Modeling and Assimilation Office: Greenbelt, MD,
12 2005; <http://gmao.gsfc.nasa.gov/systems/geos4/Bloom.pdf>.
13
14

15 (29) Jaeglé, L.; Quinn, P.K.; Bates, T.S.; Alexander, B.; Lin, J.T. Global distribution of sea salt
16 aerosols: New constraints from in situ and remote sensing observations. *Atmos. Chem. Phys.* **2011**, *11*,
17 3137-3157.
18
19

20 (30) Liu, J.; Logan, J.A.; Jones, D.B.A.; Livesey, N.J.; Megretskaia, I.; Carouge, C.; Nedelec, P.
21 Analysis of CO in the tropical troposphere using Aura satellite data and the GEOS-Chem model:
22 Insights into transport characteristics of the GEOS meteorological products. *Atmos. Chem. Phys.* **2010**,
23 *10*, 12207–12232.
24
25

26 (31) Guenther, A.; Karl, T.; Harley, P.; Wiedinmyer, C.; Palmer, P.I.; Geron, C. Estimates of global
27 terrestrial isoprene emissions using MEGAN (Model of Emissions of Gases and Aerosols from Nature).
28 *Atmos. Chem. Phys.* **2006**, *6*, 3181-3210.
29
30

31 (32) Fukui, Y.; Doskey, P.V. Air-surface exchange of nonmethane organic compounds at a grassland
32 site: Seasonal variations and stressed emissions. *J. Geophys. Res.* **1998**, *103*, 13153-13168.
33
34

35 (33) EPA NEI 2005 inventory; <http://www.epa.gov/oar/data/>.
36
37

38 (34) Liu, H.Y.; Jacob, D.J.; Bey, I.; Yantosca, R.M. Constraints from Pb-210 and Be-7 on wet
39 deposition and transport in a global three-dimensional chemical tracer model driven by assimilated
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1 meteorological fields. *J. Geophys. Res.* **2001**, *106*, 12109-12128.

2
3 (35) Mari, C.; Jacob, D.J.; Bechtold, P. Transport and scavenging of soluble gases in a deep
4 convective cloud. *J. Geophys. Res.* **2000**, *105*, 22255-22267.

5
6
7
8
9 (36) Wang, Y.H.; Jacob, D.J.; Logan, J.A. Global simulation of tropospheric O₃-NO_x-hydrocarbon
10 chemistry: 1. Model formulation. *J. Geophys. Res.* **1998**, *103*, 10713-10725.

11
12
13
14
15 (37) Wesely, M.L. Parameterization of surface resistances to gaseous dry deposition in regional-scale
16 numerical models. *Atmos. Environ.* **1989**, *23*, 1293-1304.

17
18
19
20
21 (38) Rodgers, C.D., *Inverse methods for atmospheric sounding: Theory and practice*. World
22 Scientific: Singapore, 2000.

23
24
25
26
27 (39) Tarantola, A., *Inverse Problem Theory: Methods for Data Fitting and Model Parameter*
28 *Estimation*. Elsevier: New York, 1987.

29
30
31
32 (40) Stavrakou, T.; Guenther, A.; Razavi, A.; Clarisse, L.; Clerbaux, C.; Coheur, P.F.; Hurtmans, D.;
33 Karagulian, F.; De Mazière, M.; Vigouroux, C.; Amelynck, C.; Schoon, N.; Laffineur, Q.; Heinesch, B.;
34 Aubinet, M.; Rinsland, C.; Müller, J.F. First space-based derivation of the global atmospheric methanol
35 emission fluxes. *Atmos. Chem. Phys.* **2011**, *11*, 4873-4898.

36
37
38
39
40
41
42 (41) Hafner, S.D.; Montes, F.; Rotz, C.A.; Mitloehner, F. Ethanol emission from loose corn silage and
43 exposed silage particles. *Atmos. Environ.* *44*, 4172-4180.

44
45
46
47
48 (42) Malkina, I.L.; Kumar, A.; Green, P.G.; Mitloehner, F.M. Identification and Quantitation of
49 Volatile Organic Compounds Emitted from Dairy Silages and Other Feedstuffs. *Journal of*
50 *Environmental Quality* *40*, 28-36.

51
52
53
54
55
56 (43) Arellano, A.F.; Hess, P.G. Sensitivity of top-down estimates of CO sources to GCTM transport.
57
58
59
60

1 *Geophys. Res. Lett.* **2006**, *33*, L21807, doi:10.1029/2006GL027371.

2
3 (44) Heald, C.L.; Jacob, D.J.; Jones, D.B.A.; Palmer, P.I.; Logan, J.A.; Streets, D.G.; Sachse, G.W.;
4 Gille, J.C.; Hoffman, R.N.; Nehr Korn, T. Comparative inverse analysis of satellite (MOPITT) and
5 aircraft (TRACE-P) observations to estimate Asian sources of carbon monoxide. *J. Geophys. Res.* **2004**,
6 *109*, D23306, doi:10.1029/2004JD005185.
7
8

9
10
11
12 (45) Kopacz, M.; Jacob, D.; Henze, D.K.; Heald, C.L.; Streets, D.G.; Zhang, Q. A comparison of
13 analytical and adjoint Bayesian inversion methods for constraining Asian sources of CO using satellite
14 (MOPITT) measurements of CO columns. *J. Geophys. Res.* **2009**, *114*, D04305,
15 doi:10.1029/2007JD009264.
16
17
18

19 (46) Peylin, P.; Baker, D.; Sarmiento, J.; Ciais, P.; Bousquet, P. Influence of transport uncertainty on
20 annual mean and seasonal inversions of atmospheric CO₂ data. *J. Geophys. Res.* **2002**, *107*, 4385,
21 doi:10.1029/2001JD000857.
22
23
24

25 (47) Karl, T.; Harley, P.; Emmons, L.; Thornton, B.; Guenther, A.; Basu, C.; Turnipseed, A.; Jardine,
26 K. Efficient atmospheric cleansing of oxidized organic trace gases by vegetation. *Science* **2010**, *330*,
27 816-819.
28
29
30

31 (48) Hu, L.; Millet, D.B.; Mohr, M.J.; Wells, K.C.; Griffis, T.J.; Helmig, D. Sources and seasonality
32 of atmospheric methanol based on tall tower measurements in the US Upper Midwest. *Atmos. Chem.*
33 *Phys.* **2011**, *11*, 11145-11156.
34
35
36

37 (49) Wells, K.C.; Millet, D.B.; Hu, L.; Cady-Pereira, K.E.; Xiao, Y.; Shephard, M.W.; Clerbaux, C.L.;
38 Clarisse, L.; Coheur, P.-F.; Apel, E.C.; Gouw, J.d.; Warneke, C.; Singh, H.B.; Goldstein, A.H.; Sive,
39 B.C. Tropospheric methanol observations from space: Retrieval evaluation and constraints on the
40 seasonality of biogenic emissions. *Atmos. Chem. Phys.* **2011**, *submitted*.
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1 (50) Giebel, B.M.; Swart, P.K.; Riemer, D.D. New insights to the use of ethanol in automotive fuels:
2 A stable isotopic tracer for fossil- and bio-fuel combustion inputs to the atmosphere. *Environ. Sci.*
3 *Technol.* **2011**, *45*, 6661-6669.
4
5
6
7

8 (51) Hill, J.; Polasky, S.; Nelson, E.; Tilman, D.; Huo, H.; Ludwig, L.; Neumann, J.; Zheng, H.C.;
9 Bonta, D. Climate change and health costs of air emissions from biofuels and gasoline. *P. Natl. Acad.*
10 *Sci. USA* **2009**, *106*, 2077-2082.
11
12
13
14
15

16 (52) Jacobson, M.Z. Effects of ethanol (E85) versus gasoline vehicles on cancer and mortality in the
17 United States. *Environ. Sci. Technol.* **2007**, *41*, 4150-4157.
18
19
20
21

22 (53) US Energy Information Administration; <http://www.eia.gov/>.
23
24

25 (54) US Department of Energy Transportation Energy Data Book;
26 http://cta.ornl.gov/data/appendix_b.shtml.
27
28
29
30

31 (55) Yanowitz, J.; McCormick, R.L. Effect of E85 on tailpipe emissions from light-duty vehicles. *J.*
32 *Air & Waste Manage. Assoc.* **2009**, *59*, 172-182.
33
34
35
36

37 (56) *Regulatory Impact Analysis: Renewable Fuel Standard Program*; EPA420-R-07-004; US
38 Environmental Protection Agency (EPA): Washington, DC, 2007;
39 <http://www.epa.gov/otaq/renewablefuels/420r07004-sections.htm>.
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

Table 1. North American ethanol sources: A priori forward model emissions versus a posteriori emissions optimized on the basis of aircraft measurements.

Simulation	Optimized Regions ¹				US Total ²	North American Total	Cost Function Reduction
	Western Region	Central Region	Eastern Region	Mexico			
<u>Ethanol Emissions (GgC/y)</u>							
A Priori	148	243	370	208	613	969	-
Opt1 ³	227	146	174	124	440	671	1.52
<u>Sensitivity Runs: Deviation From Best Guess Optimization Opt1</u>							
Opt2-DEP ⁴	+4%	+8%	+4%	+4%	+5%	+5%	1.44
Opt3-OH ⁵	-3%	+14%	+5%	-5%	+4%	+2%	1.47
Opt4-MET ⁶	+6%	+43%	-13%	+17%	+6%	+11%	1.42
Opt5-SPR ⁷	+17%	NA	NA	-5%	NA	NA	1.66
Opt6-SUM ⁸	-12%	-6%	+1%	NA	-6%	-17%	1.58

¹Regions are defined as shown in Fig. S3. The Western, Central and Eastern regions include the US and Canada. ²Continental US. ³Best-guess optimization. ⁴Optimization using increased dry deposition loss rate. ⁵Optimization using modified OH concentrations (see text). ⁶Optimization using GEOS-4 meteorological fields. ⁷Optimization using springtime aircraft data only. ⁸Optimization using summertime aircraft data only.

1
2
3 **Figure 1.** Airborne ethanol observations. Shown are flight tracks from the Intercontinental Transport
4 Experiment, Phase A (INTEX-A: DC8 aircraft, Jul–Aug 2004) and Phase B (INTEX-B: DC8 and C130
5 aircraft, Apr–May 2006); and the Megacity Initiative: Local And Global Research Observations
6 (MILAGRO: DC8 and C130 aircraft, Mar 2006). Flight tracks are plotted over boundary layer ethanol
7 concentrations ($P > 750$ hPa, Mar–Aug mean) from the GEOS-Chem a priori simulation.
8
9
10
11
12
13
14
15
16
17

18 **Figure 2.** Difference plot showing ethanol concentrations in the boundary layer from the GEOS-Chem a
19 priori simulation minus the corresponding measured values from the aircraft campaigns. Model values
20 are sampled along the flight track at the time of the observations.
21
22
23
24
25
26
27
28

29 **Figure 3.** Seasonal ethanol emissions from North America. Emissions for present-day (based on the
30 Opt1 source optimization) are compared to those for the EISA and All-E85 scenarios as described in the
31 text. Numbers inset give the total North American ethanol source (Canada + US + Mexico).
32
33
34
35
36
37
38

39 **Figure 4.** Projected change (%) in atmospheric ethanol, acetaldehyde, PAN:NO_y ratio, PAN, and NO_x
40 over and downwind of North America for a US transition to ethanol-based fuel (All-E85 scenario). The
41 largest relative increases for ethanol, acetaldehyde and PAN occur during fall, winter and spring
42 because of substantial biogenic emissions of ethanol and other VOCs during summer. NO_x emissions
43 are lower for E85-fueled vehicles, but the associated concentration decrease downwind of North
44 America is offset by increased long-range transport of PAN.
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

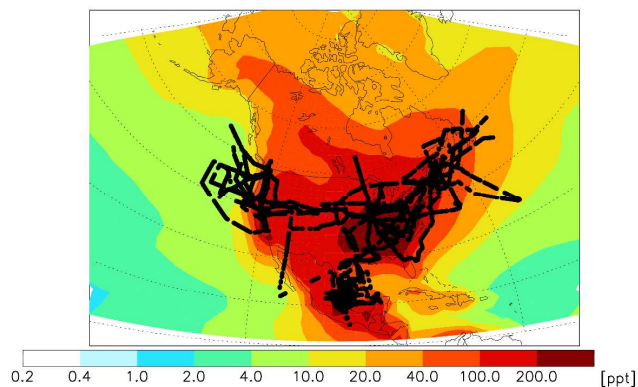


Figure 1. Airborne ethanol observations. Shown are flight tracks from the Intercontinental Transport Experiment, Phase A (INTEX-A: DC8 aircraft, Jul–Aug 2004) and Phase B (INTEX-B: DC8 and C130 aircraft, Apr–May 2006); and the Megacity Initiative: Local And Global Research Observations (MILAGRO: DC8 and C130 aircraft, Mar 2006). Flight tracks are plotted over boundary layer ethanol concentrations ($P > 750$ hPa, Mar–Aug mean) from the GEOS-Chem a priori simulation.

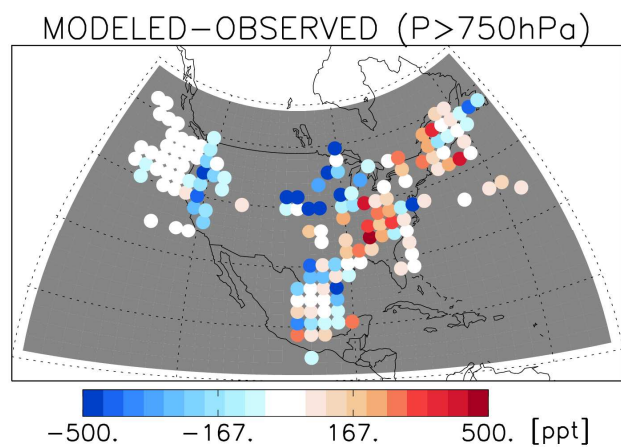


Figure 2. Difference plot showing ethanol concentrations in the boundary layer from the GEOS-Chem a priori simulation minus the corresponding measured values from the aircraft campaigns. Model values are sampled along the flight track at the time of the observations.

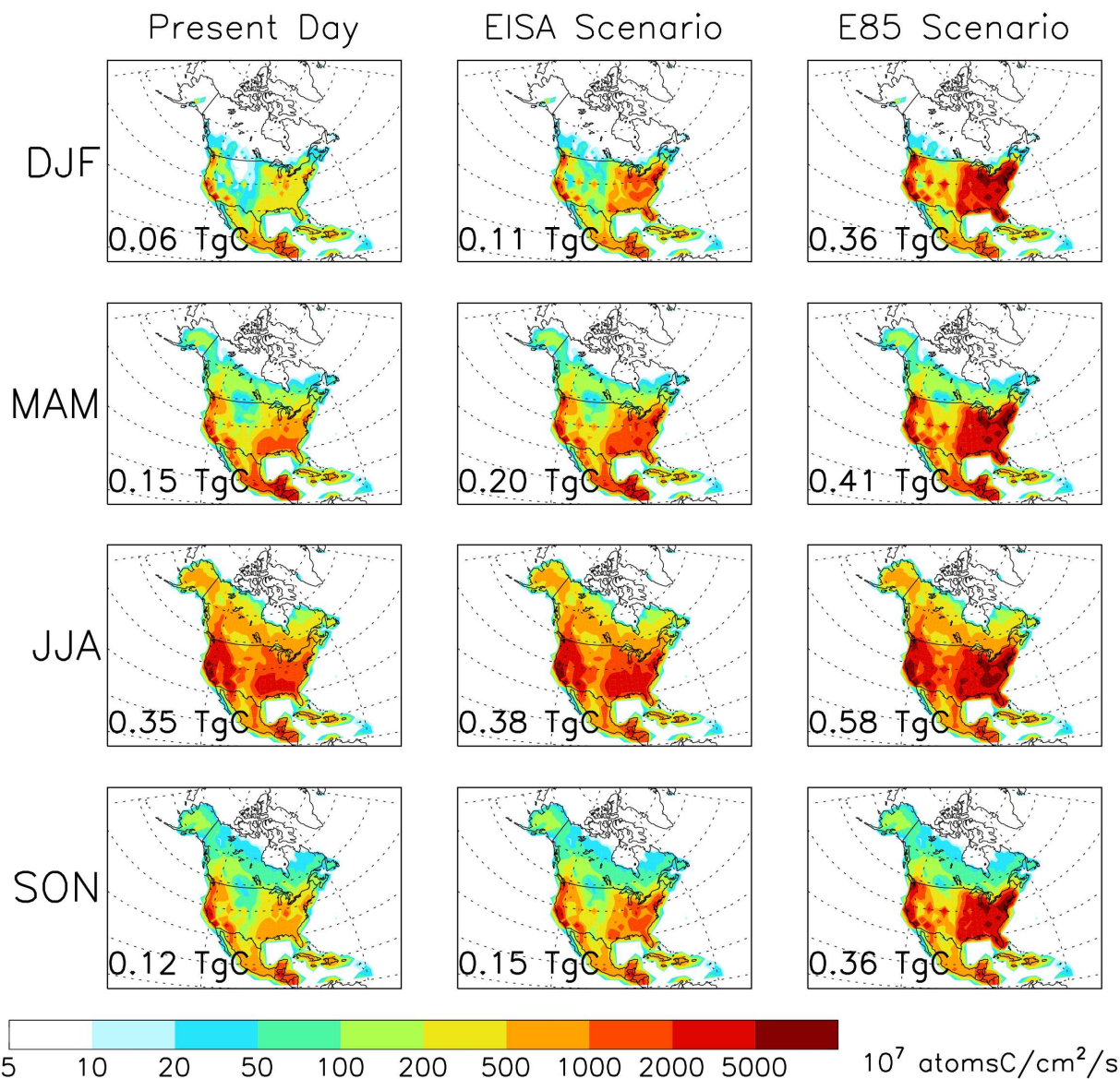


Figure 3. Seasonal ethanol emissions from North America. Emissions for present-day (based on the Opt1 source optimization) are compared to those for the EISA and All-E85 scenarios as described in the text. Numbers inset give the total North American ethanol source (Canada + US + Mexico).

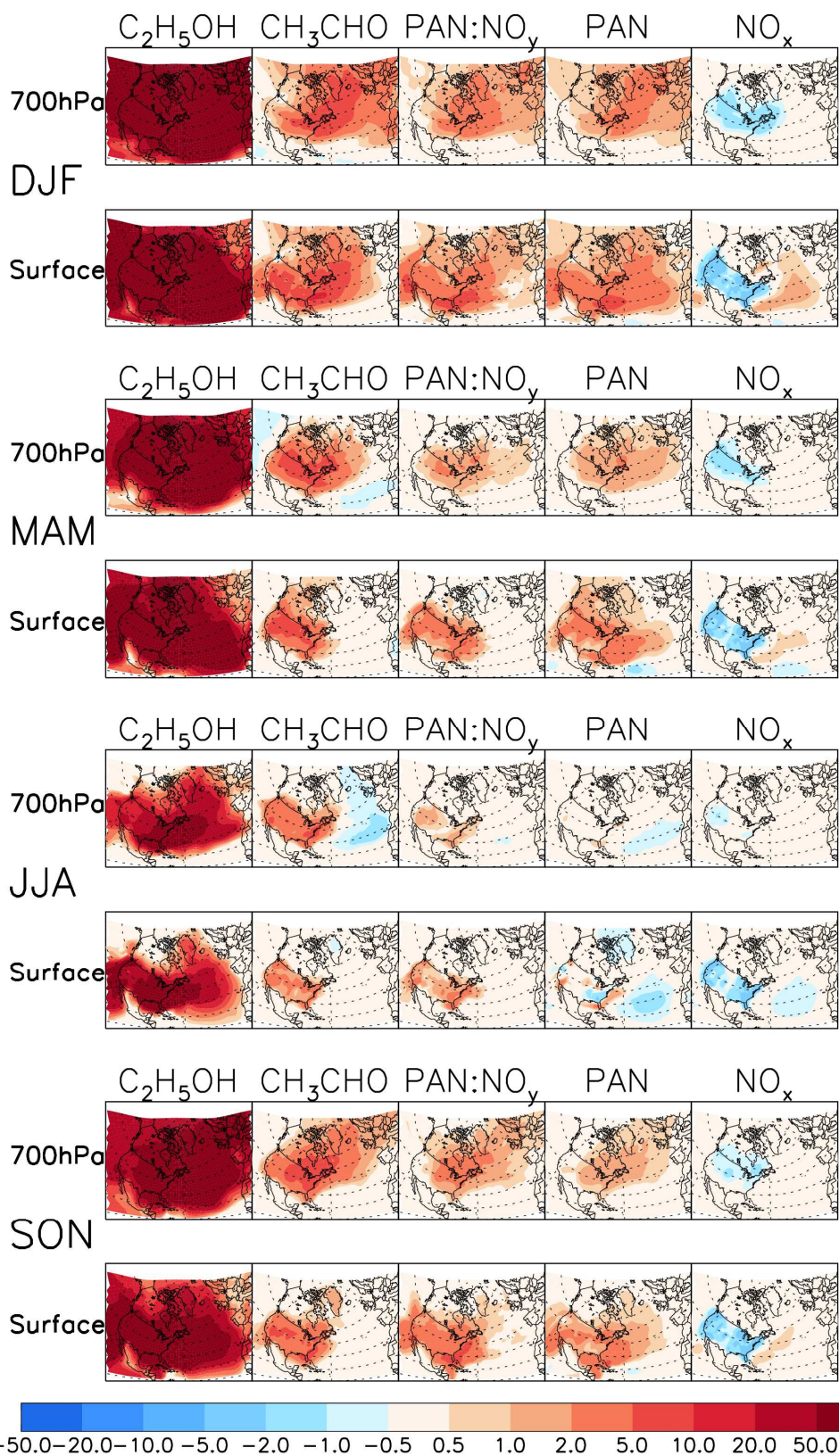
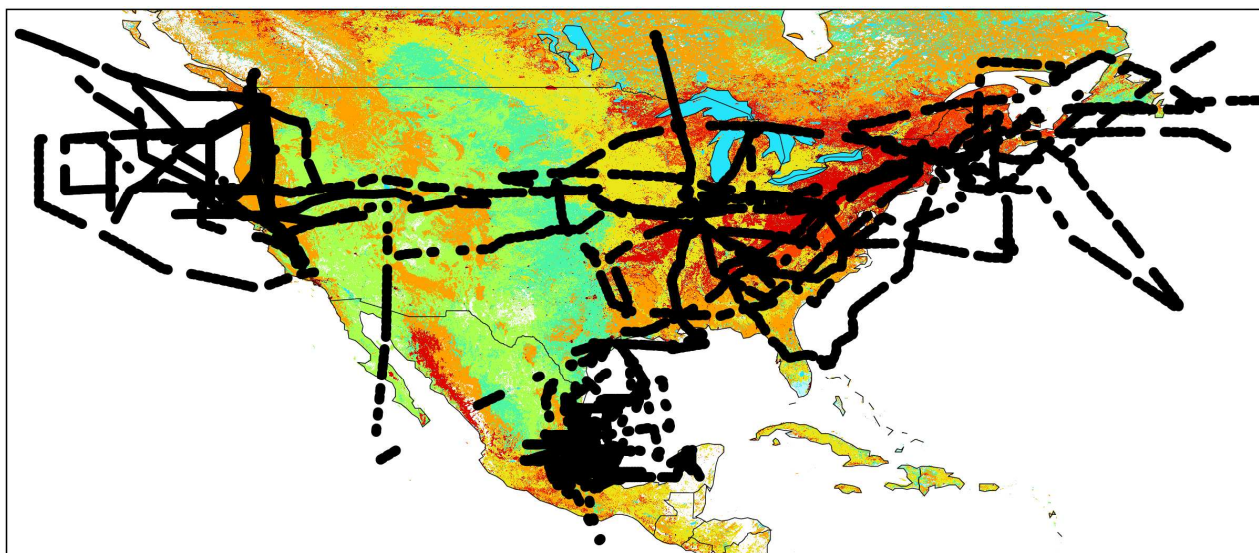


Figure 4. Projected change (%) in atmospheric ethanol, acetaldehyde, PAN: NO_y ratio, PAN, and NO_x

1 over and downwind of North America for a US transition to ethanol-based fuel (All-E85 scenario). The
2 largest relative increases for ethanol, acetaldehyde and PAN occur during fall, winter and spring
3 because of substantial biogenic emissions of ethanol and other VOCs during summer. NO_x emissions
4 are lower for E85-fueled vehicles, but the associated concentration decrease downwind of North
5 America is offset by increased long-range transport of PAN.
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60



TOC Art.